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# Instructions for Contributors to PLANT AND CELL PHYSIOLOGY

(Revised December 1968)

1. Contributors to PLANT AND CELL PHYSIOLOGY are required to hold membership in the Japanese Society of Plant Physiologists.

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- ( iv ) Text
- ( v ) Acknowledgements
- ( vi ) References
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6. **Texts** should be prepared (except short communications), as far as possible, under the headings of Introduction, Materials and methods, Results and Discussion. Consult recent issues of PLANT AND CELL PHYSIOLOGY for paragraph headings and subheadings.

7. **Tables** should be prepared in a form consistent with recent issues of PLANT AND CELL PHYSIOLOGY. Vertical rules are not permitted. Tables should be numbered consecutively with Arabic numerals. Explanatory material should be given in table footnotes rather than in the heading. The footnotes should be referred to by superscript figures: *a*, *b*, *c*,... Each table should be written on a separate page.

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- (2) Vidaver, W. and C.S. French. Oxygen uptake and evolution following monochromatic flashes in Ulva and an action spectrum for system I. ibid., 40, 7-12 (1965).
- (3) Kornberg, A. Adenosine phosphokinase. In Methods in Enzymology 2. Edited by S.P. Colowick and N.O. Kaplan. p. 497-500. Academic Press Inc., New York, N. Y. (1965).

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Manuscripts should be sent to: Dr. Atsushi Takimoto, Laboratory of Applied Botany, Faculty of Agriculture, Kyoto University, Kyoto, Japan.

#### Abbreviations commonly used in PLANT AND CELL PHYSIOLOGY

5'-diphosphates of adenosine, cytidine, guanosine, inosine, uridine	ADP, CDP, GDP, IDP, UDP
adenosine 5'-phosphate, etc.	AMP, etc.
adenosine 5'-triphosphate, etc.	ATP, etc.
ampere	amp
centigrade	°C
centimeter(s)	cm
coenzyme A and its acyl derivatives	CoA and acyl-CoA
concentration (in tables)	conc
counts per minute	cpm
Curie	Ci
deoxyribonuclease	DNase
deoxyribonucleic acid	DNA
ethylenediaminetetraacetate	EDTA
Figure(s)	Fig.
flavin adenine dinucleotide and its reduced form	FAD and FADH <sub>2</sub>
flavin mononucleotide	FMN
gram(s)	g
gravity	× g

hour(s)	hr
kilogram(s)	kg
liter(s)	spell out
meter(s)	m
Michaelis constant	$K_m$
microcurie(s)	$\mu\text{Ci}$
microgram(s)	$\mu\text{g}$
microliter(s)	$\mu\text{l}$
micromolar(s) (unit of conc, $\mu$ mole/liter)	$\mu\text{M}$
micromole(s) (unit of mass)	$\mu\text{mole(s)}$
microvolt(s)	$\mu\text{v}$
microwatt(s)	$\mu\text{w}$
millicurie(s)	mCi
milligram(s)	mg
milliliter(s)	ml
millimeter(s)	mm
millimicron(s)	$\text{m}\mu$
millimolar (unit of conc, mmole/liter)	mM
millimole(s) (unit of mass)	mmole(s)
millivolt(s)	mv
minute(s)	min
molar (mole/liter)	M
mole(s) (a gram molecule)	spell out
nicotinamide adenine dinucleotide and its reduced form	NAD and NADH <sub>2</sub> (or NAD <sup>+</sup> and NADH + H <sup>+</sup> )
nicotinamide adenine dinucleotide phosphate and its reduced form	NADP and NADPH <sub>2</sub> (or NADP <sup>+</sup> and NADPH + H <sup>+</sup> )
normal (conc)	N
normal (in trivial names in organic compounds)	<i>n</i> -
number	No.
optical density (optical density at 260 $\text{m}\mu$ )	OD (OD <sub>260</sub> )
orthophosphate	P <sub>i</sub>
part(s) per million	ppm
percent	%
precipitate (in tables)	ppt
pyrophosphate	PP <sub>i</sub>
revolutions per minute	rpm
ribonuclease	RNase
ribonucleic acid	RNA
second(s)	sec
supernatant (in tables)	sup
tris (hydroxymethyl) aminomethane	Tris
ultraviolet	UV
volt(s)	v
volume/volume (conc)	v/v
watt(s)	w
weight/volume (conc)	w/v
weight/weight (conc)	w/w
For isotopes, use <sup>14</sup> C, <sup>35</sup> S instead of C <sup>14</sup> , S <sup>35</sup> .	

The metric system is adopted as standard: If English measures are to be used for any reason, metric equivalents must be supplied in parentheses.

